# Structure of Dicarbonyliodo( $\boldsymbol{\eta}^{\mathbf{5}}$-pentamethylcyclopentadienyl)iridium(III) Tetrafluoroborate $\left[\mathbf{C p}{ }^{*} \operatorname{Ir}(\mathbf{C O})_{2}{ }^{1}\right] \mathrm{BF}_{4}$ 

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#### Abstract

Ir}\left(\mathrm{C}_{10} \mathrm{H}_{15}\right)(\mathrm{I})(\mathrm{CO})_{2}\right]\left[\mathrm{BF}_{4}\right], \quad M_{r}=597 \cdot 17\), orthorhombic, $P c a 2_{1}, a=13.035$ (4), $b=10.517$ (3), $c=12.013(4) \AA, \quad V=1646.9 \AA^{3}, \quad Z=4, \quad D_{x}=$ $2.41 \mathrm{~g} \mathrm{~cm}^{-3}, \quad \lambda(\mathrm{Mo} K \alpha)=0.71069 \AA, \quad \mu=$ $99.74 \mathrm{~cm}^{-1}, F(000)=1095.75, T=296 \mathrm{~K}, R=0.044$, $w R=0.049$ for 1014 observed reflections. The Ir atom in the cation is coordinated to an $\eta^{5}$ pentamethylcyclopentadienyl group, two carbonyl groups and an iodine. Selected bond distances and angles are: $\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid $1 \cdot 88$, $\mathrm{Ir}-\mathrm{C}(1)$ 1.89 (3), Ir-C(2) 1.81 (4), Ir-I 2.717 (3) $\AA$; C(1)-$\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid 126, $\mathrm{C}(2)-\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid 132, $\mathrm{I}-\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid $120, \mathrm{C}(1)-\mathrm{Ir}-\mathrm{C}(2)$ 92 (2), C(1)-Ir--I 90 (2), C(2)-Ir-I 84 (1).


Experimental. Compound isolated from reaction of $\mathrm{Cp}{ }^{*} \operatorname{Re}(\mathrm{CO})_{2} \mathrm{I}_{2}$ and $\mathrm{Cp}{ }^{*} \mathrm{Ir}(\mathrm{CO})_{2}\left(\mathrm{Cp}^{*}=\eta^{5}-\mathrm{C}_{5} \mathrm{Me}_{5}\right)$ with $\mathrm{AgBF}_{4}$ in ethanol; crystals obtained from $\mathrm{CH}_{2} \mathrm{Cl}_{2}$-hexane (Zhang \& Sutton, 1990). A palegreen crystal $0.39 \times 0.24 \times 0.12 \mathrm{~mm}$ was mounted on a glass fibre. Intensity data were collected at 296 (1) K with an Enraf-Nonius CAD-4F diffractometer using graphite-monochromatized Mo $K \alpha$

Table 1. Positional parameters and $B_{\mathrm{iso}} / B_{\text {eq }}$ values $\left(\AA^{2}\right)$ for $\left[\mathrm{Cp}{ }^{*} \mathrm{Ir}(\mathrm{CO})_{2} \mathrm{I}\right] \mathrm{BF}_{4}$

|  | $x$ | $y$ | $z$ | $B_{\text {iso }} / B_{\text {eq }}{ }^{\dagger}$ |
| :---: | :---: | :---: | :---: | :---: |
| Ir | 0.03319 (7) | $0 \cdot 12921$ (9) | 0 | 2.59 |
| I | $0 \cdot 1188$ (3) | 0.0189 (3) | $0 \cdot 1822$ (3) | 6.41 |
| C(1) | -0.0778 (27) | 0.0139 (35) | 0.009 (6) | $6 \cdot 1$ (8) |
| O(1) | -0.1450 (20) | -0.049 (3) | 0.024 (3) | 7.3 (7) |
| $\mathrm{C}(2)$ | $0 \cdot 1107$ (32) | 0.0155 (42) | -0.076 (3) | 5.5 (9) |
| O(2) | 0.1582 (24) | -0.040 (3) | -0.138 (3) | 7.4 (8) |
| C(11) | -0.0471 (26) | 0.3055 (32) | 0.056 (3) | $2 \cdot 8$ (7) |
| C(12) | 0.0537 (28) | $0 \cdot 3198$ (36) | 0.075 (3) | $4 \cdot 1$ (8) |
| C(13) | 0.1162 (20) | 0.3131 (24) | -0.015 (3) | $3 \cdot 1$ (6) |
| C(14) | 0.0507 (24) | $0 \cdot 3033$ (30) | -0.105 (3) | $2 \cdot 6$ (6) |
| C(15) | -0.0557 (32) | $0 \cdot 2929$ (40) | -0.060 (4) | 2.7 (10) |
| C(21) | -0.1388 (38) | 0.3207 (42) | $0 \cdot 125$ (4) | 6.0 (11) |
| C(22) | 0.0987 (26) | 0.3523 (31) | 0.198 (3) | 4.7 (7) |
| C(23) | 0.2321 (24) | 0.3356 (27) | -0.020 (3) | 4.4 (7) |
| C(24) | 0.0765 (26) | 0.3000 (34) | -0.226 (3) | 4.6 (7) |
| C(25) | -0.1483 (31) | 0.2948 (35) | -0.133 (3) | $3 \cdot 2$ (8) |
| $\mathrm{F}(1)$ | 0.4618 (16) | 0.4148 (18) | 1.0242 (26) | 6.7 (5) |
| F (2) | 0.5226 (20) | 0.2298 (24) | 0.9615 (20) | 7.8 (7) |
| F(3) | 0.6156 (25) | 0.3548 (29) | 1.0728 (27) | $9 \cdot 1$ (8) |
| F(4) | 0.5956 (20) | 0.4077 (24) | 0.9024 (22) | $2 \cdot 9$ (7) |
| B | 0.5439 (24) | 0.347 (3) | 0.986 (3) | 2.9 (7) |

$\dagger B_{\text {eq }}$ is the mean of the principal axes of the thermal ellipsoid.
radiation. Lattice parameters were determined from 25 reflections ( $15 \leq \theta \leq 18^{\circ}$ ). 1678 independent reflections were measured ( $3 \leq 2 \theta \leq 50^{\circ} ; h: 0 \rightarrow 15 ; k$ : $0 \rightarrow 12 ; l: 0 \rightarrow 14)$ using $\omega-2 \theta$ scans; 1014 observed reflections $\left[I_{o} \geq 2.5 \sigma\left(I_{o}\right)\right]$; scan range ( $1 \cdot 10+$ $0.35 \tan \theta)^{\circ}$; scan speed $0 \cdot 19-5.49^{\circ} \mathrm{min}^{-1}$; intensities of three standards $(0 \overline{4} 2,121,2 \overline{1} \overline{1})$ were measured every 80 min of acquisition time and showed no long term change and had an r.m.s. deviation of $<1.4 \%$; data reduction included intensity scaling, Lorentz and polarization corrections and an analytical absorption correction, $\psi$-scan checked, transmission $0 \cdot 240-0.545$ (Alcock, 1969). Structure solved by Patterson synthesis, and refined by full-matrix least squares and Fourier synthesis methods. The assumption that the space group is non-centrosymmetric $P c a 2_{1}$ was confirmed by the successful refinement. The positions of some methyl-group H atoms were located in difference Fourier maps $(\sin \theta / \lambda<$ $0 \cdot 3 \AA^{-1}$ ) and were used to calculate remaining methyl H -atom positions $[d(\mathrm{C}-\mathrm{H})=0.95 \AA$ ]. Final refinement included anisotropic thermal parameters for $\operatorname{Ir}$ and I , and isotropic thermal parameters for

Table 2. Selected bond lengths $(\AA)$ and bond angles $\left({ }^{\circ}\right)$ for $\left[\mathrm{Cp} * \operatorname{Ir}(\mathrm{CO})_{2} \mathrm{I}\right] \mathrm{BF}_{4}$

| $\mathrm{Ir}-\mathrm{I}$ |  | $\mathrm{C}(11)-\mathrm{C}(12) \quad 1.3$ | 1.34 (5) |
| :---: | :---: | :---: | :---: |
| $\mathrm{Ir}-\mathrm{C}(1) \quad 1.89$ (3) |  | $\mathrm{C}(12)-\mathrm{C}(13) \quad 1.3$ | 1.35 (5) |
| $\mathrm{Ir}-\mathrm{C}(2) \quad 1.81$ (4) |  | $\mathrm{C}(13)-\mathrm{C}(14) \quad 1$. | 1.39 (4) |
| $\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid 1.88 |  | $\mathrm{C}(14)-\mathrm{C}(15) \quad 1$. | 1.49 (5) |
| $\mathrm{Ir}-\mathrm{C}(11) \quad 2.23$ (3) |  | $\mathrm{C}(15)-\mathrm{C}(11) \quad 1$. | 1.41 (6) |
| $\mathrm{Ir}-\mathrm{C}(12) \quad 2.21$ (4) |  | $\mathrm{C}(11)-\mathrm{C}(21) \quad 1$. | 1.46 (6) |
| $\mathrm{Ir}-\mathrm{C}(13) \quad 2.22$ (3) |  | $\mathrm{C}(12)-\mathrm{C}(22) \quad 1$. | 1.63 (5) |
| $\mathrm{Ir}-\mathrm{C}(14) \quad 2.23$ (3) |  | $\mathrm{C}(13)-\mathrm{C}(23) \quad 1$. | 1.53 (4) |
| $\mathrm{Ir}-\mathrm{C}(15) \quad 2.20$ (4) |  | $\mathrm{C}(14)-\mathrm{C}(24) \quad 1$. | 1.49 (5) |
| $\mathrm{C}(1)-\mathrm{O}(1) \quad 1 \cdot 11$ (5) |  | $\mathrm{C}(15)-\mathrm{C}(25) \quad 1$. | 1.49 (6) |
| $\mathrm{C}(2)-\mathrm{O}(2) \quad 1.14$ (5) |  |  |  |
| $\mathrm{I}-\mathrm{Ir}-\mathrm{C}(1)$ | 90 (2) | $\mathrm{C}(15)-\mathrm{C}(14)-\mathrm{C}(13)$ | ) 107 (3) |
| $\mathrm{I}-\mathrm{Ir}-\mathrm{C}(2)$ | 84 (1) | $\mathrm{C}(12)-\mathrm{C}(11)-\mathrm{C}(21)$ | 134 (4) |
| $\mathrm{C}(1)-\mathrm{Ir}-\mathrm{C}(2)$ | 92 (2) | $\mathrm{C}(15)-\mathrm{C}(11)-\mathrm{C}(21)$ | 1) 120 (3) |
| $\mathrm{Ir}-\mathrm{C}(1)-\mathrm{O}(1)$ | 173 (5) | $\mathrm{C}(11)-\mathrm{C}(12)-\mathrm{C}(22)$ | ) 122 (3) |
| $\mathrm{Ir}-\mathrm{C}(2)-\mathrm{O}(2)$ | 168 (4) | $\mathrm{C}(13)-\mathrm{C}(12)-\mathrm{C}(22)$ | ) 121 (3) |
| $\mathrm{C}(1)-\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid | 126 | $\mathrm{C}(12)-\mathrm{C}(13)-\mathrm{C}(23)$ | ) 128 (3) |
| $\mathrm{C}(2)-\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid | 132 | $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(23)$ | ) 126 (3) |
| $\mathrm{I}-\mathrm{Ir}-\mathrm{C}_{5} \mathrm{Me}_{5}$ centroid | 120 | $\mathrm{C}(13)-\mathrm{C}(14)-\mathrm{C}(24)$ | 4) 129 (3) |
| $\mathrm{C}(11)-\mathrm{C}(15)-\mathrm{C}(14)$ | 106 (3) | $\mathrm{C}(15)-\mathrm{C}(14)-\mathrm{C}(24)$ | 4) 124 (3) |
| $\mathrm{C}(12)-\mathrm{C}(11)-\mathrm{C}(15)$ | 105 (3) | $\mathrm{C}(14)-\mathrm{C}(15)-\mathrm{C}(25)$ | ) 123 (4) |
| $\mathrm{C}(13)-\mathrm{C}(12)-\mathrm{C}(11)$ | 117 (3) | $\mathrm{C}(11)-\mathrm{C}(15)-\mathrm{C}(25)$ | ) 130 (4) |
| $\mathrm{C}(14)-\mathrm{C}(13)-\mathrm{C}(12)$ | 105 (3) |  |  |

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other non-H atoms. H atoms were included in fixed positions in structure factor calculations. Refinement was considered complete when the shift/e.s.d. ratio was less than 0.02 ; 94 parameters varied. Final residuals $R=0.044, w R=0.049$; goodness of fit 1.004 ; quantity minimized $\sum w\left(F_{o}-F_{c}\right)^{2}$, where $w=1$. Highest peak in final electron density map was $3.4(2)$ e $\AA^{-3}$ at a distance $1.01 \AA$ from Ir atom and $\rho_{\min }$ was $-1.3(2)$ e $\AA^{-3}$. Analytical forms of scattering factors for neutral atoms used (International Tables for X-ray Crystallography, 1974, Vol. IV, Tables 2.2B and 2.3.1.); all non-H-atom scattering factors corrected for real and imaginary components of anomalous dispersion. Inspection of $w \Delta^{2}$ as a function of $\sin \theta / \lambda, F_{o}$ and values of $h, k, l$ showed no unusual features or trends. Positional parameters are listed in Table 1, and bond distances and angles are listed in Table $2 . \dagger$ Programs used were from the NRC VAX Crystal Structure System (Gabe, Le Page, Charland, Lee \& White, 1989) including their version of ORTEP (Johnson, 1965), and were run on a MicroVAX II computer. Fig. 1. shows the cation with the atom-labelling scheme.

Related literature. The structures of the related compounds [Cp $\left.{ }^{*} \mathrm{Ir}\left(\mathrm{PMe}_{3}\right)_{2} \mathrm{Cl}\right]\left[\mathrm{PF}_{6}\right]$ (Kaner, Kouvetakis \& Mayorga, 1986) and [ $\left.\mathrm{Cp}{ }^{*} \operatorname{Ir}(\mathrm{CO})_{2}\left(\mathrm{CH}_{2} \mathrm{Cl}\right)\right]-$ $\left[\mathrm{Re}_{2}(\mathrm{CO})_{6}(\mu-\mathrm{Cl})_{x}\left(\mu-\mathrm{Br}_{3}-x\right)\right]$ (Einstein, Glavina, Pomeroy \& Willis, 1986) have been reported.

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Fig. 1. View of the cation $\left[\mathrm{Cp} * \operatorname{Ir}(\mathrm{CO})_{2}\right]^{+}$with atom labelling.

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# Structure of $\boldsymbol{N}$-Ethyl-m-fluorophenylsuccinimide 

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#### Abstract

Ethyl-3-(3-fluorophenyl)-2,5-pyrrolidinedione, $\mathrm{C}_{12} \mathrm{H}_{12} \mathrm{FNO}_{2}, M_{r}=221 \cdot 23$, monoclinic, $P 2_{1} / c$, $a=6.623$ (1),$\quad b=10.810$ (2), $c=15 \cdot 297$ (2) $\AA, \quad \beta=$ $96.82(1)^{\circ}, \quad V=1087.6(3) \AA^{3}, \quad Z=4, \quad D_{x}=$ $1.351 \mathrm{Mg} \mathrm{m}^{-3}, \quad \lambda(\mathrm{CuK} \mathrm{\alpha})=1.54178 \AA, \quad \mu=$ $0.84 \mathrm{~mm}^{-1}, F(000)=464$, room temperature, final $R$ $=0.052$ for 1577 observed reflections (of 1969 unique data). The succinimide fragment of the molecule is non-planar and has an envelope


conformation [the deviation of Cl from the plane $\mathrm{N} 1-\mathrm{C} 4-\mathrm{C} 2-\mathrm{C} 3$ is -0.096 (2) $\AA$ and the dihedral angle between planes $\mathrm{C} 4-\mathrm{C} 1-\mathrm{C} 2$ and $\mathrm{N} 1-\mathrm{C} 4-\mathrm{C} 2-\mathrm{C} 3$ is $\left.5 \cdot 8(2)^{\circ}\right]$. The phenyl ring is planar. The angle between the best planes of the five-membered ring and the phenyl ring is $87.1(1)^{\circ}$.

Experimental. The title compound recrystallized from ethanol and gave colourless crystals. Crystal
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[^0]:    $\dagger$ Lists of structure factors, anisotropic thermal parameters and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54100 ( 8 pp .). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

