## Structure of Dicarbonyliodo( $\eta^5$ -pentamethylcyclopentadienyl)iridium(III) Tetrafluoroborate [Cp\*Ir(CO)<sub>2</sub>I]BF<sub>4</sub>

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(Received 19 November 1990; accepted 19 March 1991)

Abstract.  $[Ir(C_{10}H_{15})(I)(CO)_2][BF_4], M_r = 597.17,$ orthorhombic,  $Pca2_1$ , a = 13.035 (4), b = 10.517 (3), c = 12.013 (4) Å, V = 1646.9 Å<sup>3</sup>, Z = 4,  $D_x =$  $2.41 \text{ g cm}^{-3}$  $\lambda(\text{Mo } K\alpha) = 0.71069 \text{ Å},$  $\mu =$  $99.74 \text{ cm}^{-1}$ , F(000) = 1095.75, T = 296 K, R = 0.044, wR = 0.049 for 1014 observed reflections. The Ir atom in the cation is coordinated to an  $\eta^{5}$ pentamethylcyclopentadienyl group, two carbonyl groups and an iodine. Selected bond distances and angles are:  $Ir - C_5 Me_5$  centroid 1.88, Ir - C(1)1.89 (3), Ir-C(2) 1.81 (4), Ir-I 2.717 (3) Å; C(1)-Ir— $C_5Me_5$  centroid 126, C(2)—Ir— $C_5Me_5$  centroid 132, I—Ir— $C_5Me_5$  centroid 120, C(1)—Ir—C(2) 92 (2), C(1)—Ir--I 90 (2), C(2)—Ir--I 84 (1)°.

**Experimental.** Compound isolated from reaction of  $Cp^*Re(CO)_2I_2$  and  $Cp^*Ir(CO)_2$  ( $Cp^* = \eta^5-C_5Me_5$ ) with AgBF<sub>4</sub> in ethanol; crystals obtained from  $CH_2Cl_2$ -hexane (Zhang & Sutton, 1990). A palegreen crystal  $0.39 \times 0.24 \times 0.12$  mm was mounted on a glass fibre. Intensity data were collected at 296 (1) K with an Enraf-Nonius CAD-4F diffractometer using graphite-monochromatized Mo  $K\alpha$ 

radiation. Lattice parameters were determined from 25 reflections  $(15 \le \theta \le 18^\circ)$ . 1678 independent reflections were measured ( $3 \le 2\theta \le 50^\circ$ ; h:  $0 \rightarrow 15$ ; k:  $0 \rightarrow 12$ ; l:  $0 \rightarrow 14$ ) using  $\omega - 2\theta$  scans; 1014 observed reflections  $[I_o \ge 2.5\sigma(I_o)]$ ; scan range (1.10 + $0.35\tan\theta$ )°; scan speed  $0.19-5.49^{\circ}$  min<sup>-1</sup>; intensities of three standards  $(0\overline{4}2, 121, 2\overline{1}\overline{1})$  were measured every 80 min of acquisition time and showed no long term change and had an r.m.s. deviation of < 1.4%; data reduction included intensity scaling, Lorentz and polarization corrections and an analytical absorption correction,  $\psi$ -scan checked, transmission 0.240-0.545 (Alcock, 1969). Structure solved by Patterson synthesis, and refined by full-matrix least squares and Fourier synthesis methods. The assumption that the space group is non-centrosymmetric  $Pca2_1$  was confirmed by the successful refinement. The positions of some methyl-group H atoms were located in difference Fourier maps  $(\sin\theta/\lambda <$  $0.3 \text{ Å}^{-1}$ ) and were used to calculate remaining methyl H-atom positions [d(C-H) = 0.95 Å]. Final refinement included anisotropic thermal parameters for Ir and I, and isotropic thermal parameters for

Table 1. Positional parameters and  $B_{iso}/B_{eq}$  values (Å<sup>2</sup>) for [Cp\*Ir(CO)<sub>2</sub>I]BF<sub>4</sub>

Table 2.	Selected bond lengths (Å) and bond angles (°)									
for [Cp*Ir(CO) <sub>2</sub> I]BF <sub>4</sub>										

					T. T	2 717 (2)		C(11) $C(12)$	1.24 (5)
	x	v	Ζ	$B_{\rm iso}/B_{\rm co}^{\dagger}$		2.717 (3)		C(1) - C(12)	1.34 (3)
Ir	0.03319(7)	0.12921(9)	0	2.59	Ir - C(1)	1.89 (3)		C(12) - C(13)	1.35 (5)
ï	0.1188(3)	0.0189(3)	0.1822 (3)	6.41	Ir - C(2)	1.81 (4)		C(13) - C(14)	1.39 (4)
c (1)	-0.0778(27)	0.0139(35)	0.009 (6)	6.1 (8)	Ir-C <sub>5</sub> Me <sub>5</sub> centrol	11.88		C(14) - C(15)	1.49 (5)
	-0.1450(20)	-0.049(3)	0.024(3)	7.3(7)	lr - C(11)	2.23 (3)		C(15) - C(11)	1.41 (6)
C(2)	0.1107(32)	0.0155(42)	-0.076(3)	5.5 (9)	Ir-C(12)	2.21 (4)		C(11) - C(21)	1.46 (6)
O(2)	0.1582(24)	-0.040(3)	-0.138(3)	7.4 (8)	Ir—C(13)	2.22 (3)		C(12) - C(22)	1.63 (5)
C(11)	-0.0471(24)	0.3055 (32)	0.056(3)	2.8(7)	IrC(14)	2.23 (3)		C(13) - C(23)	1.53 (4)
C(11)	-0.0471 (20)	0.2108 (32)	0.075 (3)	$\frac{20}{11}(8)$	Ir—C(15)	2·20 (4)		C(14) - C(24)	1.49 (5)
C(12)	0.0337(28)	0.2121 (24)	0.015(3)	2.1 (6)	C(1)—O(1)	1.11 (5)		C(15)C(25)	1.49 (6)
C(13)	0.1102(20)	0.3131(24)	-0.013(3)	3.1 (0)	C(2)—O(2)	1.14 (5)			
C(14)	0.0507 (24)	0.3033(30)	-0.105(3)	2.0 (0)					
C(15)	-0.0557(32)	0.2929(40)	-0.060 (4)	2.7 (10)	I - Ir - C(1)		90 (2)	C(15)-C(14)-C	(13) 107 (3)
C(21)	-0.1388(38)	0.3207(42)	0.125 (4)	6.0 (11)	I - Ir - C(2)		84 (1)	C(12) - C(11) - C	(21) 134 (4)
C(22)	0.0987 (26)	0.3523 (31)	0.198 (3)	4.7 (7)	C(1) - Ir - C(2)		92 (2)	C(15) - C(11) - C	(21) 120 $(3)$
C(23)	0.2321 (24)	0.3356 (27)	<i>−</i> 0·020 (3)	4·4 (7)	Ir - C(1) - O(1)		173 (5)	C(11) - C(12) - C	(22) 122 $(3)$
C(24)	0.0765 (26)	0.3000 (34)	-0·226 (3)	4·6 (7)	Ir = C(2) = O(2)		168 (4)	C(13) - C(12) - C	(22) 121 (3)
C(25)	- 0·1483 (31)	0.2948 (35)	-0.133 (3)	3.2 (8)	C(1) Ir $C(2)$	rentroid	126	C(12) - C(13) - C(13	(23) 128 (3)
F(1)	0.4618 (16)	0.4148 (18)	1.0242 (26)	6.7 (5)	C(2) Ir $C(Me)$	centroid	120	C(12) = C(13) = C(13	(23) 126 $(3)$
F(2)	0.5226 (20)	0.2298 (24)	0.9615 (20)	7.8 (7)	$C(2) = \Pi = C_{3} M c_{3}$	roid	120	C(13) - C(14) - C(14	(23) 120 $(3)$
F(3)	0.6156 (25)	0.3548 (29)	1.0728 (27)	9.1 (8)	C(11) = C(15) = C(15)	14)	106 (2)	C(15) = C(14) = C	(24) 129 (3)
F(4)	0.5956 (20)	0.4077 (24)	0.9024 (22)	2.9 (7)	C(1) - C(1) - C(1)	14)	100(3)	C(13) - C(14) - C(14	(24) 124 $(3)$
B	0.5439 (24)	0.347 (3)	0.986 (3)	2.9 (7)	C(12) - C(11) - C(12)	15)	105 (3)	C(14) - C(15) - C	(25) 125 (4) (25) 120 (4)
-	· · · · · · · · · · · · · · · · · · ·		(- /		C(13) - C(12) - C(12)	11)	117(3)	-U(1)-U(1)-U(1)-U(1)	(25) 130 (4)

C(14) - C(13) - C(12)

 $\dagger B_{ca}$  is the mean of the principal axes of the thermal ellipsoid.

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105 (3)

other non-H atoms. H atoms were included in fixed positions in structure factor calculations. Refinement was considered complete when the shift/e.s.d. ratio was less than 0.02; 94 parameters varied. Final residuals R = 0.044, wR = 0.049; goodness of fit 1.004; quantity minimized  $\sum w(F_o - F_c)^2$ , where w = 1. Highest peak in final electron density map was 3.4(2) e Å<sup>-3</sup> at a distance 1.01 Å from Ir atom and  $\rho_{\min}$  was -1.3 (2) e Å<sup>-3</sup>. Analytical forms of scattering factors for neutral atoms used (International Tables for X-ray Crystallography, 1974, Vol. IV, Tables 2.2B and 2.3.1.); all non-H-atom scattering factors corrected for real and imaginary components of anomalous dispersion. Inspection of  $w\Delta^2$  as a function of  $\sin\theta/\lambda$ ,  $F_{a}$  and values of h, k, l showed no unusual features or trends. Positional parameters are listed in Table 1, and bond distances and angles are listed in Table 2.† Programs used were from the NRC VAX Crystal Structure System (Gabe, Le Page, Charland, Lee & White, 1989) including their version of ORTEP (Johnson, 1965), and were run on a MicroVAX II computer. Fig. 1. shows the cation with the atom-labelling scheme.

**Related literature.** The structures of the related compounds  $[Cp*Ir(PMe_3)_2Cl][PF_6]$  (Kaner, Kouvetakis & Mayorga, 1986) and  $[Cp*Ir(CO)_2(CH_2Cl)]$ - $[Re_2(CO)_6(\mu-Cl)_x(\mu-Br_{3-x})]$  (Einstein, Glavina, Pomeroy & Willis, 1986) have been reported.

<sup>†</sup> Lists of structure factors, anisotropic thermal parameters and H-atom parameters have been deposited with the British Library Document Supply Centre as Supplementary Publication No. SUP 54100 (8 pp.). Copies may be obtained through The Technical Editor, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.



Fig. 1. View of the cation  $[Cp*Ir(CO)_2I]^+$  with atom labelling.

This work was supported by NSERC Canada through operating and infrastructure grants to FWBE and DS. We thank Johnson Matthey Co. for the generous loan of iridium trichloride.

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Acta Cryst. (1991). C47, 1978-1980

## Structure of N-Ethyl-m-fluorophenylsuccinimide

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(Received 31 July 1990; accepted 15 January 1991)

Abstract. 1-Ethyl-3-(3-fluorophenyl)-2,5-pyrrolidinedione, C<sub>12</sub>H<sub>12</sub>FNO<sub>2</sub>,  $M_r = 221 \cdot 23$ , monoclinic,  $P2_1/c$ ,  $a = 6 \cdot 623$  (1),  $b = 10 \cdot 810$  (2),  $c = 15 \cdot 297$  (2) Å,  $\beta =$  $96 \cdot 82$  (1)°,  $V = 1087 \cdot 6$  (3) Å<sup>3</sup>, Z = 4,  $D_x =$  $1 \cdot 351$  Mg m<sup>-3</sup>,  $\lambda$ (Cu  $K\alpha$ ) =  $1 \cdot 54178$  Å,  $\mu =$  $0 \cdot 84$  mm<sup>-1</sup>, F(000) = 464, room temperature, final R $= 0 \cdot 052$  for 1577 observed reflections (of 1969 unique data). The succinimide fragment of the molecule is non-planar and has an envelope conformation [the deviation of C1 from the plane N1-C4-C2-C3 is -0.096(2) Å and the dihedral angle between planes C4-C1-C2 and N1-C4-C2-C3 is  $5.8(2)^{\circ}$ ]. The phenyl ring is planar. The angle between the best planes of the five-membered ring and the phenyl ring is  $87.1(1)^{\circ}$ .

**Experimental.** The title compound recrystallized from ethanol and gave colourless crystals. Crystal

0108-2701/91/091978-03\$03.00

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